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Laser Annealing Processes in Semiconductor Technology: Theory, Modeling and Applications in Nanoelectronics synthesizes the scientific and technological advances of laser annealing processes for current and emerging nanotechnologies. The book provides an overview of the laser-matter interactions of materials and recent advances in modeling of laser-related phenomena, with the bulk of the book focusing on current and emerging (beyond-CMOS) applications. Reviewed applications include laser annealing of CMOS, group IV semiconductors, superconducting materials, photonic materials, 2D materials. This comprehensive book is ideal for post-graduate students, new entrants, and experienced researchers in academia, research and development in materials science, physics and engineering. Introduces the fundamentals of laser materials and device fabrication methods, including laser-matter interactions and laser-related phenomena Addresses advances in physical modeling and in predictive simulations of laser annealing processes such as atomistic modeling and TCAD simulations Reviews current and emerging applications of laser annealing processes such as CMOS technology and group IV semiconductors Emphasises on contemporary applications and an intuitive problem-solving approach that helps students discover the exciting potential of chemical science. This book incorporates fresh applications from the three major areas of modern research: materials, environmental chemistry, and biological science. Chemical principles are fundamental to the Earth sciences, andgeoscience students increasingly require a firm grasp of basicchemistry to succeed in their studies. The enlarged third editionof this highly regarded textbook introduces the student to such‘geo-relevant’ chemistry, presented in the same lucidand accessible style as earlier editions, but the new edition hasbeen strengthened in its coverage of environmental geoscience andincorporates a new chapter introducing isotope geochemistry. The book comprises three broad sections. The first (Chapters1–4) deals with the basic physical chemistry of geologicalprocesses. The second (Chapters 5–8) introduces thewave-mechanical view of the atom and explains the various types ofchemical bonding that give Earth materials their diverse anddistinctive properties. The final chapters (9–11) survey thegeologically relevant elements and isotopes, and explain theirformation and their abundances in the cosmos and the Earth. Thebook concludes with an extensive glossary of terms; appendicescover basic maths, explain basic solution chemistry, and list thechemical elements and the symbols, units and constants used in thebook. This volume presents a sound foundation for understanding abstract concepts (physical properties such as fugacity, or chemical processes, such as distillation) of phase and reaction equilibria, and shows you how to apply these concepts to solve practical problems using numerous, clear examples. The book encourages the use of MATHCAD to write programs specific to each problem, enabling you to easily track mistakes and understand the order of magnitude of the various quantities involved. Provides guidelines in order to choose the 'best' equation of state suitable for the particular situation Includes up-to-date information, comprehensive in-depth content and current examples in each chapter Provides the right tools in order to and encourages you to use MATHCAD to write your own specific programs Includes many well organized problems (with solutions), which are extensions of the examples enabling conceptual understanding to quantitative/real problem solving Includes all mathematical background required for solving problems encountered in phase and reaction equilibria Provides a Solutions Manual (for instructors in pdf form) allowing the use of the book in advanced thermodynamic courses Explore the latest research in biopharmaceutics from leading contributors in the field In Biopharmaceutics - From Fundamentals to Industrial Practice, distinguished Scientists from the UK's Academy of Pharmaceutical Sciences Biopharmaceutica Focus Group deliver a comprehensive examination of the tools used within the field of biopharmaceutics and their applications to drug development. This edited volume is an indispensable tool for anyone seeking to better understand the field of biopharmaceutics as it rapidly develops and evolves. Beginning with an expansive introduction to the basics of biopharmaceutics and the context that underpins the field, the included resources go on to discuss how biopharmaceutics are integrated into product development within the pharmaceutical industry. Explorations of how the regulatory aspects of biopharmaceutics function, as well as the impact of physiology and anatomy on the rate and extent of drug absorption, follow. Readers will find insightful discussions of physiologically based modeling as a valuable asset in the biopharmaceutics toolkit and how to apply the principles of the field to special populations. The book goes on to discuss: Thorough introductions to biopharmaceutics, basic pharmacokinetics, and biopharmaceutics measures Comprehensive explorations of solubility, permeability, and dissolution Practical discussions of the use of biopharmaceutics to inform candidate drug selection and optimization, as well as biopharmaceutics tools for rational formulation design In-depth examinations of biopharmaceutics classification systems and regulatory biopharmaceutics, as well as regulatory biopharmaceutics and the impact of anatomy and physiology Perfect for professionals working in the pharmaceutical and biopharmaceutical industries, Biopharmaceutics - From Fundamentals to Industrial Practice is an incisive and up-to-date resource on the practical, pharmaceutical applications of the field. The chemistry, structure and biological importance of octacalcium phosphate (OCP) are thoroughly discussed in this volume, which covers a wide scope of topics relevant to the field of calcium phosphates in general. The first chapter reviews the structures of OCP and other important calcium phosphates. It is followed by a description of the mimicry of OCP growth in vivo using in vitro model systems and an extensive review of OCP in biological systems. The remarkable ability of OCP to incorporate biologically relevant organic carboxylates into its structure is also discussed, along with the solubility and surface properties of calcium phosphate salts, and amorphous calcium phosphate and its relationship to OCP. The last chapter centers on the emerging technology of calcium phosphate cements as bone defect repair materials. The book provides in-depth, up-to-date information to all scientists studying calcium phosphate chemistry, biomineralization, pathological calcification, dental caries, biomaterials, and recovery of environmental phosphorus. Of the thousands of novel compounds that a drug discovery project team invents and that bind to the therapeutic target, typically only a fraction of these have sufficient ADME/Tox properties to become a drug product.

Understanding ADME/Tox is critical for all drug researchers, owing to its increasing importance in advancing high quality candidates to clinical studies and the processes of drug discovery. If the properties are weak, the candidate will have a high risk of failure or be less desirable as a drug product. This book is a tool and resource for scientists engaged in, or preparing for, the selection and optimization process. The authors describe how properties affect in vivo pharmacological activity and impact in vitro assays. Individual drug-like properties are discussed from a practical point of view, such as solubility, permeability and metabolic stability, with regard to fundamental understanding, applications of property data in drug discovery and examples of structural modifications that have achieved improved property performance. The authors also review various methods for the screening (high throughput), diagnosis (medium throughput) and in-depth (low throughput) analysis of drug properties. * Serves as an essential working handbook aimed at scientists and students in medicinal chemistry * Provides practical, step-by-step guidance on property fundamentals, effects, structure-property relationships, and structure modification strategies * Discusses improvements in pharmacokinetics from a practical chemist's standpoint This volume is of great importance to humans and other living organisms. The study of water quality draws information from a variety of disciplines including chemistry, biology, mathematics, physics, engineering, and resource management. University training in water quality is often limited to specialized courses in engineering, ecology, and fisheries curricula. This book also offers a basic understanding of water quality to professionals who are not formally trained in the subject. The revised third edition updates and expands the discussion, and incorporates additional figures and illustrative problems. Improvements include a new chapter on basic chemistry, a more comprehensive chapter on hydrology, and an updated chapter on regulations and standards. Because it employs only first-year college-level chemistry and very basic physics, the book is well-suited as the foundation for a general introductory course in water quality. It is equally useful as a guide for self-study and an in-depth resource for general readers. This long awaited second edition of a popular textbook has a simple and direct approach to the diversity and complexity of food processing. It explains the principles of operations and illustrates them by individual processes. The new edition has been enlarged to include sections on freezing, drying, psychrometry, and a completely new section on mechanical refrigeration. All the units have been converted to SI measure. Each chapter contains unworked examples to help the student gain a grasp of the subject, and although primarily intended for the student food technologist or process engineer, this book will also be useful to technical workers in the food industry This book describes the physicochemical fundamentals and biomedical principles of drug solubility. Methods to study and predict solubility in silico and in vitro are described and the role of solubility in a medicinal chemistry and pharmaceutical industry context are discussed. Approaches to modify and control solubility of a drug during the manufacturing process and of the pharmaceutical product are essential practical aspects of this book. This book arises from a workshop organized by the American Association of Pharmaceutical Scientists entitled "Optimizing the Drug-Like Properties of Leads in Drug Discovery," which took place in Parsippany, NJ in September 2004. The workshop focused on the optimization of the drug-like properties of leads in drug discovery. The volume outlines strategies and methodologies designed to guide pharmaceutical and biotechnology companies through the drug discovery and development process. Solvent systems are integral to drug development and pharmaceutical technology. This single topic encompasses numerous allied subjects running the gamut from recrystallization solvents to biorelevant media. The goal of this contribution to the AAPS Biotechnology: Pharmaceutical Aspects series is to generate both a practical handbook as well as a reference allowing the reader to make effective decisions concerning the use of solvents and solvent systems. To this end, the monograph was created by inviting recognized experts from a number of fields to author relevant sections. Specifically, 15 chapters have been designed covering the theoretical background of solubility, the effect of ionic equilibria and pH on solubilization, the use of solvents to effect drug substance crystallization and polymorph selection, the use of solvent systems in high throughput screening and early discovery, solvent use in preformulation, the use of solvents in bio-relevant dissolution and permeation experiments, solvents and their use as toxicology vehicles, solubilizing media and excipients in oral and parenteral formulation development, specialized vehicles for protein formulation and solvent systems for topical and pulmonary drug administration. The chapters are organized such that useful decision trees are included together with the scientific underpinning for their application. In addition, trends in the use of solvent systems and a balance of current views make this monograph useful to both the novice and experienced researcher and to scientists at all developmental stages from early discovery to late pharmaceutical operations. "This comprehensive ebook covers all the aspects of ADME/PK modeling including solubility, absorption, formulation, metabolic stability, drug-drug interaction potential and a special delivery tool of drug candidates. The book provides an integrated view of" Sample Text Phase equilibrium knowledge is required for the design of all sorts of chemical processes that may involve separations, reactions, fluids flow, particle micronization, etc. Indeed, different phase behavior scenarios are required for a rational conceptual process design. The aim of this chapter is to present the possible fluid mixture phase behavior that can be found in binary, ternary, and multicomponent systems. Moreover, representation of phase behavior in terms of phase diagrams is discussed. Dealing with phase diagrams of complex mixtures is not an easy task for beginners; however, very simple concepts are behind the rules for their construction. Phase diagrams are essential tools for phase equilibrium engineering as they provide valuable hints to understand the process and to assess the feasible and optimum operating regions. In this chapter, the “phenomenological” meaning of each phase behavior and its relation with molecular properties is discussed. A special attention is given to binary system phase behavior. Even though, in practice we rarely found such simple mixtures, they furnish a great deal of information for the understanding of multicomponent systems. Concepts and Experimental Protocols of Modelling and Informatics in Drug Design discusses each experimental protocol utilized in the field of bioinformatics, focusing especially on computer modeling for drug development. It helps the user in understanding the field of computer-aided molecular modeling (CAMM) by presenting solved exercises and examples. The book discusses topics such as fundamentals of molecular modeling, QSAR model generation, protein databases and how to use them to select and analyze protein structure, and pharmacophore modeling for drug targets. Additionally, it discusses data retrieval system, molecular surfaces, and freeware and online servers. The book is a valuable source for graduate students and

researchers on bioinformatics, molecular modeling, biotechnology and several members of biomedical field who need to understand more about computer-aided molecular modeling. Presents exercises with solutions to aid readers in validating their own protocol Brings a thorough interpretation of results of each exercise to help readers compare them to their own study Explains each parameter utilized in the algorithms to help readers understand and manipulate various features of molecules and target protein to design their study This book describes the physicochemical fundamentals and biomedical principles of drug solubility. Methods to study and predict solubility in silico and in vitro are described and the role of solubility in a medicinal chemistry and pharmaceutical industry context are discussed. Approaches to modify and control solubility of a drug during the manufacturing process and of the pharmaceutical product are essential practical aspects of this book. * Guidelines are provided on the reliability of various methods, as well as information for selecting the appropriate technique. * Unique coverage of the whole range of solubility measurements. * Very useful for investigators interested in embarking upon solubility measurements. Measurement of equilibrium state; Measurement of equilibrium constants; Mathematical methods used in eqwulibrium calculations; Strong acids and bases; Weak monopratic acids and bases; Precipitation and the Solubility product; Polypratec acids; Introduction to complex cormation equilibria; Complex formation; advanced topics; Oxidation-reduction equilibria; Nonideality corrections. The revised second edition updates and expands the discussion, and incorporates additional figures and illustrative problems. Improvements include a new chapter on basic chemistry, a more comprehensive chapter on hydrology, and an updated chapter on regulations and standards. This book presents the basic aspects of water quality, emphasizing physical, chemical, and biological factors. The study of water quality draws information from a variety of disciplines including chemistry, biology, mathematics, physics, engineering, and resource management. University training in water quality is often limited to specialized courses in engineering, ecology, and fisheries curricula. This book also offers a basic understanding of water quality to professionals who are not formally trained in the subject. Because it employs only first-year college-level chemistry and very basic physics, the book is well-suited as the foundation for a general introductory course in water quality. It is equally useful as a guide for self-study and an in-depth resource for general readers. This book provides a modern and easy-to-understand introduction to the chemical equilibria in solutions. It focuses on aqueous solutions, but also addresses non-aqueous solutions, covering acid–base, complex, precipitation and redox equilibria. The theory behind these and the resulting knowledge for experimental work build the foundations of analytical chemistry. They are also of essential importance for all solution reactions in environmental chemistry, biochemistry and geochemistry as well as pharmaceutics and medicine. Each chapter and section highlights the main aspects, providing examples in separate boxes. Questions and answers are included to facilitate understanding, while the numerous literature references allow students to easily expand their studies. The role of specific molecular interactions in influencing the solubility behavior of organic compounds are examined, particularly the role of hydrogen bonding. Shows how specific interactions can be used to elicit preferential solubility. Emphasizes interactions occurring in environments of low polarity and explains and predicts solubility phenomena in self-associated solvents. Also considers the kinetics of diffusion and dissolution. Food Engineering Handbook, Two-Volume Set provides a stimulating and up-to-date review of food engineering phenomena. It also addresses the basic and applied principles of food engineering methods used in food processing operations around the world. Combining theory with a practical, hands-on approach, this set examines the thermophysical properties and modeling of selected processes such as chilling, freezing, and dehydration, and covers the key aspects of food engineering, from mass and heat transfer to steam and boilers, heat exchangers, diffusion, and absorption. Comprised of Food Engineering Handbook: Food Engineering Fundamentals and Food Engineering Handbook: Food Process Engineering, this comprehensive resource: Explains the interactions between different food constituents that might lead to changes in food properties Describes the characterization of the heating behavior of foods, their heat transfer, heat exchangers, and the equipment used in each food engineering method Discusses rheology, fluid flow, evaporation, distillation, size reduction, mixing, emulsion, and encapsulation Provides case studies of solid–liquid and supercritical fluid extraction and food behaviors Explores fermentation, enzymes, fluidized-bed drying, and more Presenting cutting-edge information on new and emerging food engineering processes, Food Engineering Handbook, Two-Volume Set offers a complete reference on the fundamental concepts, modeling, quality, safety, and technologies associated with food engineering and processing operations today. * The present work is designed to provide a practical introduction to aqueous equilibrium phenomena for both students and research workers in chemistry, biochemistry, geochemistry, and interdisciplin ary environmental fields. The pedagogical strategy I have adopted makes heavy use of detailed examples of problem solving from real cases arising both in laboratory research and in the study of systems occurring in nature. The procedure starts with mathematically complete equations that will provide valid solutions of equilibrium problems, instead of the traditional approach through approximate concentrations and idealized, infinite-dilution assumptions. There is repeated emphasis on the use of corrected, conditional equilibrium constants and on the checking of numerical results by substitution in complete equations and/or against graphs of species distributions. Graphical methods of calculation and display are used extensively because of their value in clarifying equilibria and in leading one quickly to valid numerical approximations. The coverage of solution equilibrium phenomena is not, however, exhaustively comprehensive. Rather, I have chosen to offer funda mental and rigorous examinations of homogeneous step-equilibria and their interactions with solubility and redox equilibria. Many examples are worked out in detail to demonstrate the use of equi librium calculations and diagrams in various fields of investigation. Aqueous solubility is one of the major challenges in the early stages of drug discovery. One of the most common and effective methods for enhancing solubility is the addition of an organic solvent to the aqueous solution. Along with an introduction to cosolvency models, the Handbook of Solubility Data for Pharmaceuticals provides an extensive database of solubility for pharmaceuticals in mono solvents and binary solvents. Aqueous solubility data can be found in the Handbook of Aqueous Solubility Data by Samuel Yalkowsky and Yan He. Visit www.crcpress.com for more information. In addition to the experimental efforts to measure the solubility of drugs in mono and mixed solvents, this book discusses the advantages and limitations of a number of mathematical models used to predict the solubility in mono or

mixed solvent systems. It covers the pharmaceutical cosolvents and other organic solvents that are used in syntheses, separations, and other pharmaceutical processes. The solutes featured include the available data for official drugs, drug candidates, precursors of drugs, metabolites, and degradation products of pharmaceuticals. The author also presents the solubilities of amino acids since they play an important role in peptide drug properties. Collecting drug solubilities in various cosolvents, this time-saving handbook includes the mixtures and model constants needed to predict undetermined solubilities. It describes mathematical models that enable data to be derived and provides estimates on how drugs are likely to behave in a given cosolvent. A software program and associated user manual are available on the author’s website. New directions in supercritical fluids science and technology, fluorescence spectroscopy studies of intermolecular interactions in supercritical fluids, solvation structure in supercritical fluid mixtures based on molecular distribution functions, gibbs-ensemble Monte Carlo simulations of phase equilibria in supercritical fluid mixtures, spectroscopic determination of solvent strength and structure in supercritical fluid mixtures, partition coefficients of polyethyle glycols in super critical carbon dioxide, experimental measurement of supercritical fluid-liquid phase equilibrium, vapor-liquid equilibria of fatty acid esters in supercritical fluids, four-phase equilibrium of two ternary organic systems with carbon dioxide, direct viscosity enhancement of carbon dioxide, inverse emulsion polymerization of acrylamide, interaction of polymers with near-critical carbon dioxide, fundamental kinetics of methanol oxidation in supercritical fluids, thermodynamic analysis of corrosion of iron alloys in supercritical water, electrochemical measurements of corrosion of iron alloys in supercritical water, phase and reaction equilibria considerations in the evaluation and operation of supercritical fluid reaction processes, kinetic model for supercritical delignification of wood, gas antisolvent recrystallization solids formation after the expansion of supercritical mixtures, food, pharmaceutical, and environmental applications, design of commercial plant. Reservoir Formation Damage, Third Edition, provides the latest information on the economic problems that can occur during various phases of oil and gas recovery from subsurface reservoirs, including production, drilling, hydraulic fracturing, and workover operations. The text helps readers better understand the processes causing formation damage and the factors that can lead to reduced flow efficiency in near-wellbore formation during the various phases of oil and gas production. The third edition in the series provides the most all-encompassing volume to date, adding new material on conformance and water control, hydraulic fracturing, special procedures for unconventional reservoirs, field applications design, and cost assessment for damage control measures and strategies. Understand relevant formation damage processes by laboratory and field testing Develop theories and mathematical expressions for description of the fundamental mechanisms and processes Predict and simulate the consequences and scenarios of the various types of formation damage processes encountered in petroleum reservoirs Develop methodologies and optimal strategies for formation damage control and remediation Gas Solubilities: Widespread Applications discusses several topics concerning the various applications of gas solubilities. The first chapter of the book reviews Henr's law, while the second chapter covers the effect of temperature on gas solubility. The third chapter discusses the various gases used by Horiuti, and the following chapters evaluate the data on sulfur dioxide, chlorine data, and solubility data for hydrogen sulfide. Chapter 7 concerns itself with solubility of radon, thoron, and actinon. Chapter 8 tackles the solubilities of diborane and the gaseous hydrides of groups IV, V, and VI of the periodic table. Chapter 9 discusses the solubility of gases containing fluorine, while Chapter 10 talks about Hildebrand's theory in the light of all gas solubility data. Chapter 11 covers the hydrogen halide system, while Chapter 12 deals with the solubility of gases in water and aqueous solutions of slats, inorganic acids and bases, and organic compounds. Chapter 13 discusses gases in sea water, while Chapter 14 covers aerosol propellants and Chapter 15 tackles the solubility of nitric oxide. Chapter 16 discusses the biotechnological aspects, and Chapter 17 talks about more on making holes. Chapter 18 covers the evaluation of data on phosphine. The book would be of great interest to researchers and professionals concerned with applications of the soluble nature of gases. Properties and Formulation: From Theory to Real-World Application Scientists have attributed more than 40 percent of the failures in new drug development to poor biopharmaceutical properties, particularly water insolubility. Issues surrounding water insolubility can postpone or completely derail important new drug development. Even the much-needed reformulation of currently marketed products can be significantly affected by these challenges. More recently it was reported that the percentage increased to 90% for the candidates of new chemical entities in the discovery stage and 75% for compounds under development. In the most comprehensive resource on the topic, this third edition of Water-Insoluble Drug Formulation brings together a distinguished team of experts to provide the scientific background and step-by-step guidance needed to deal with solubility issues in drug development. Twenty-three chapters systematically describe the detailed discussion on solubility theories, solubility prediction models, the aspects of preformulation, biopharmaceutics, pharmacokinetics, regulatory, and discovery support of water-insoluble drugs to various techniques used in developing delivery systems for water-insoluble drugs. This book includes more than 15 water-insoluble drug delivery systems or technologies, illustrated with case studies and featuring oral and parenteral applications. Highlighting the most current information and data available, this seminal volume reflects the significant progress that has been made in nearly all aspects of this field. The aim of this book is to provide a handy reference for pharmaceutical scientists in the handling of formulation issues related to water-insoluble drugs. In addition, this book may be useful to pharmacy and chemistry undergraduate students and pharmaceutical and biopharmaceutical graduate students to enhance their knowledge in the techniques of drug solubilization and dissolution enhancement. An Introduction to Aqueous Electrolyte Solutions is a comprehensive coverage of the subject including the development of key concepts and theory that focus on the physical rather than the mathematical aspects. Important links are made between the study of electrolyte solutions and other branches of chemistry, biology, and biochemistry, making it a useful cross-reference tool for students studying this important area of electrochemistry. Carefully developed throughout, each chapter includes intended learning outcomes and worked problems and examples to encourage student understanding of this multidisciplinary subject. * a comprehensive introduction to aqueous electrolyte solutions including the development of key concepts and theories * emphasises the connection between observable macroscopic experimental properties and interpretations made at the molecular

level * key developments in concepts and theory explained in a descriptive manner to encourage student understanding * includes worked problems and examples throughout An invaluable text for students taking courses in chemistry and chemical engineering, this book will also be useful for biology, biochemistry and biophysics students required to study electrochemistry. Aquacultural, oceanographic, and fisheries engineering, as well as other disciplines, require gas solubility data to compute the equilibrium concentration. These calculations, for example, can affect the output of aquacultural production or assist in environmental consulting. Until now, published solubility information has not been available in a consistent and uniform manner in one location. This book presents solubility concentrations of major atmospheric gases (oxygen, nitrogen, argon, carbon dioxide), noble gases (helium, neon, krypton, xenon), and trace gases (hydrogen, methane, nitrous oxide) as a function of temperature, salinity, pressure, and gas composition in a variety of formats. Data, equations, and theory are explained so that the user is able to understand the calculations and problems. Furthermore, data and solubility information are presented in a range of units to make them accessible across disciplines. This book will help the reader to look at a problem from a quantitative viewpoint and better understand carbonate chemistry. Revised from the earlier edition to include more accurate carbon dioxide tables and separate sections on the solubility of noble gases, trace gases, and oxygen in brines to provide a single resource for gas solubility data. This book is essential for all students and practitioners working in aquatic fields. A single source for highly accurate and comprehensive tables for gas solubility in aquatic systems Information provided in tables, equations, and computer programmes Theory is presented to better understand the equations and calculations A celebrated classic in the field updated and expanded to includethe latest computerized calculation techniques In 1964, James N. Butler published a book in which he presentedsome simple graphical methods of performing acid-base, solubility,and complex formation equilibrium calculations. Today, both thebook and these methods have become standard for generations ofstudents and professionals in fields ranging from environmentalscience to analytical chemistry. Named a "Citation Classic" by theScience Citation Index in 1990, the book, Ionic Equilibrium,continues to be one of the most widely used texts on the subject.So why tamper with near-perfection by attempting a revision of thatclassic? The reason is simple-- the recent rapid development andwide availability of personal computers. In the revised Ionic Equilibrium, Dr. Butler updates his 1964 workby abandoning the slide rule and graph paper for the PCspreadsheet. He also expands the original coverage with extensivematerial on basic principles and recent research. The first part of Ionic Equilibrium is devoted to the fundamentalsof acid-base, solubility, and complex formation equilibria. In thesecond part, the author discusses oxidation-reduction equilibria,develops the principles of carbon dioxide equilibria, presents casestudies demonstrating the ways in which carbon dioxide equilibriaare used in physiology and oceanography, and explores thepossibility of a pH scale for brines. The concluding chapter,written by David R. Cogley, gives examples of general computerprograms that are capable of performing equilibrium calculations onsystems of many components. Replete with real-world examples, details of importantcalculations, and practical problems, Ionic Equilibrium is an idealcourse text for students of environmental chemistry, engineering,or health; analytical chemistry; oceanography; geochemistry;biochemistry; physical chemistry; and clinical chemistry. It isalso a valuable working resource for professionals in those fieldsas well as industrial chemists involved with solution chemistry. A quick reference to basic science for anaesthetists, containing all the key information needed for FRCA exams. Metal ions play an important role in analytical chemistry, organometallic chemistry, bioinorganic chemistry, and materials chemistry. This book, Descriptive Inorganic Chemistry Researches of Metal Compounds, collects research articles, review articles, and tutorial description about metal compounds. To perspective contemporary researches of inorganic chemistry widely, the kinds of metal elements (typical and transition metals including rare earth; p, d, f-blocks) and compounds (molecular coordination compounds, ionic solid materials, or natural metalloenzyme) or simple substance (bulk, clusters, or alloys) to be focused are not limited. In this way, review chapters of current researches are collected in this book. Thermodynamic Properties of Nonelectrolyte Solutions reviews several of the more classical theories on the thermodynamics of nonelectrolyte solutions. Basic thermodynamic principles are discussed, along with predictive methods and molecular thermodynamics. This book is comprised of 12 chapters; the first of which introduces the reader to mathematical relationships, such as concentration variables, homogeneous functions, Euler’s theorem, exact differentials, and method of least squares. The discussion then turns to partial molar quantities, ideal and nonideal solutions, and empirical expressions for predicting the thermodynamic properties of multicomponent mixtures from binary data. The chapters that follow explore binary and ternary mixtures containing only nonspecific interactions; the thermodynamic excess properties of liquid mixtures and ternary alcohol-hydrocarbon systems; and solubility behavior of nonelectrolytes. This book concludes with a chapter describing the use of gas-liquid chromatography in determining the activity coefficients of liquid mixtures and mixed virial coefficients of gaseous mixtures. This text is intended primarily for professional chemists and researchers, and is invaluable to students in chemistry or chemical engineering who have background in physical chemistry and classical thermodynamics. This extensive reference/text explores the principles, instrumentation, processes, and programs of pharmaceutical solid science as well as new aspects on one-component systems, micromeritics, polymorphism, solid-state stability, cohesion, powder flow, blending, single-unit sustained release, and tablet coating. Reveals unique approaches in pharmaceutical solid science not previously published in any other text! Providing current data on crystallization, dissolution from particles and polydisperse populations, powder volumes and densities, comminution, wet granulation, and hard-shell capsules, Advanced Pharmaceutical Solids describes moisture isotherms with crystalline solids documents the effects of moisture on solid-state stability highlights tablet physics and principles explains sustained release by microencapsulation presents prediction equations for solubility in binary solvents discusses particle sizes and diameters identifies Brunauer, Emmett, and Teller Isotherms and more! Considering properties of solids, permeamitry and gas absorption methods, amorphates, and purification by pH-change precipitation, Advanced Pharmaceutical Solids is an essential reference for pharmacists; pharmaceutical scientists; medicinal, physical, surface, colloid, and analytical chemists and biochemists; and an effective text for upper-level undergraduate and graduate students in these disciplines.